

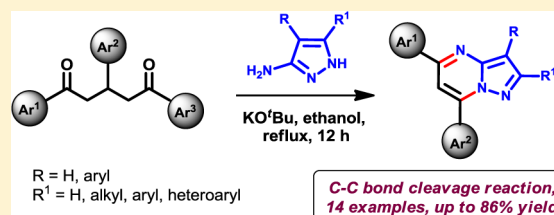
Carbon–Carbon Bond Cleavage Reaction: Synthesis of Multisubstituted Pyrazolo[1,5-*a*]pyrimidines

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S Supporting Information

ABSTRACT: A new carbon–carbon bond cleavage reaction was developed for the efficient synthesis of multisubstituted pyrazolo[1,5-*a*]pyrimidines. This base induced reaction of 1,3,5-trisubstituted pentane-1,5-diones and substituted pyrazoles afforded good yields of the pyrazolo[1,5-*a*]pyrimidines.



The cleavage of carbon–carbon bond is a significant issue in organic chemistry due to the inert nature of the C–C bond.¹ Although, the importance of this C–C bond cleavage had already resulted different methodologies for the cleavage of C–C,² C=C³ and C≡C⁴ bonds, the development of new routes for selective cleavage of C–C bond still remains as an important and challenging goal for the chemists and biologists.

The pyrazole fused heterocycle pyrazolo[1,5-*a*]pyrimidine is the key structural motif of several drugs and pesticides. For example, the hypnotic drugs zaleplon and indiplon, the anxiolytic drug ocinaplon, and the fungicide pyrazophos have this central motif of pyrazolo[1,5-*a*]pyrimidine (Figure 1). The

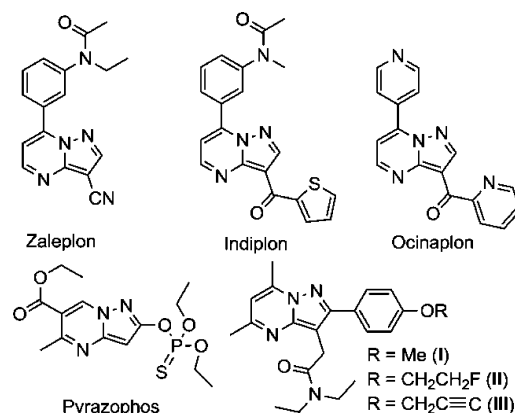


Figure 1. Examples of drugs/biologically active pyrazolo[1,5-*a*]pyrimidines.

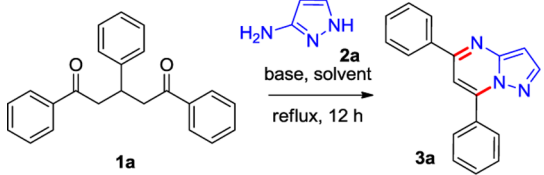
pyrazolo[1,5-*a*]pyrimidines I–III (Figure 1) are found to possess high affinity for translocator protein (TSPO), which is revealed as an attractive target in anticancer therapy.⁵ The pyrazolo[1,5-*a*]pyrimidine derivatives are also known for their wide range of biological activities such as antimicrobial, antibacterial, antitrichomonal, antischistosomal, anticancer, antitumor, etc.⁶ Some of the pyrazolo[1,5-*a*]pyrimidine derivatives are found to have potent and selective Pim-1

inhibitory activity, excellent activity against wild-type HIV-1 and CK2 Kinase inhibitory activity.⁷ The importance of these pyrazolo[1,5-*a*]pyrimidines has led to the development of new methods for the synthesis of these molecules, typically by the condensation reaction of (i) aminopyrazole with 1,2-allenyl ketones,⁸ (ii) aminopyrazole with enaminonitriles or enamionones,⁹ and (iii) aminopyrazole with 1, 3-dicarbonyl compounds or α,β -unsaturated carbonyl compounds.¹⁰ Recently, we also reported the synthesis of pyrazolo[1,5-*a*]pyrimidines by the regioselective palladium catalyzed reaction of β -halovinyl aldehydes with 3-aminopyrazoles.^{11c} In spite of the presence of methods for the synthesis of pyrazolo[1,5-*a*]pyrimidines, due to the tremendous biological profile of these heterocycles, new efficient methods which can accommodate a broad range of substituents on the heterocyclic scaffold are still highly desirable. In continuation of our work on search for novel methods for the syntheses of important heterocycles,¹¹ herein, we report an unprecedented reaction of 1,3,5-trisubstituted pentane-1,5-diones with substituted 3-amino pyrazoles in the presence of base, which proceeds via C–C bond cleavage for the easy construction of multisubstituted pyrazolo[1,5-*a*]pyrimidines. Initially, we selected 1,5-diketone **1a** and 3-amino-1*H*-pyrazole (**2a**) as the model substrates for the synthesis of compound **3a** (Table 1). Refluxing a mixture of **1a** and **2a** in anhydrous ethanol in the presence of 2 equiv of NaOMe for 12 h furnished pyrazolo[1,5-*a*]pyrimidine **3a** in 57% yield (entry 1, Table 1). The product **3a** was identified from ¹H NMR, ¹³C NMR and mass spectral data. To determine the ideal base and solvent for this base induced reaction, we studied this model reaction with some other bases and solvents as shown in Table 1. The bases such as KOMe, NaH and KOH led to lower yield of **3a** (entries 2 and 7–8). Gratifyingly, when we used NaO^tBu and KO^tBu as the bases (2 equiv) in the above reaction, the reaction afforded 78% and 84% yields of **3a**, respectively, in ethanol (entries 3–4). Further studies of KO^tBu

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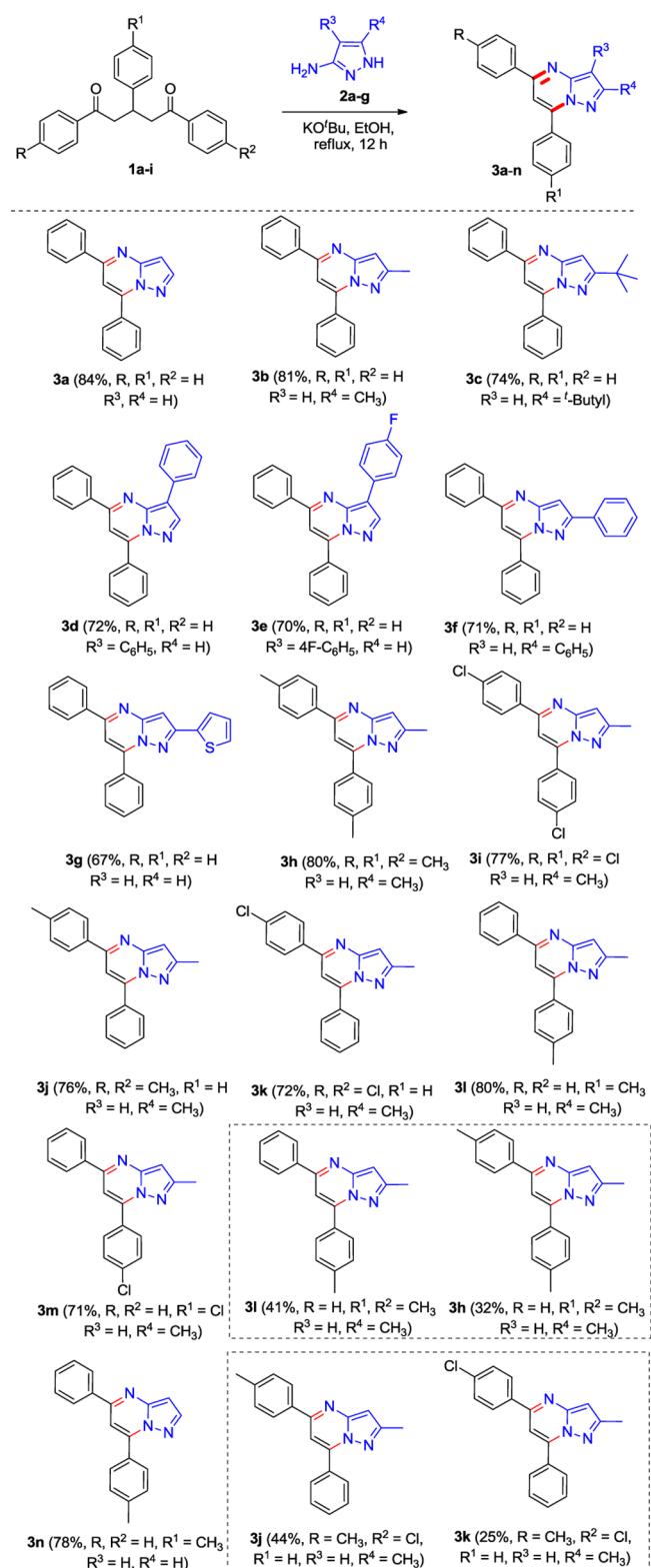
Table 1. Optimization of the Reaction Conditions for 3a



entry	base ^a	solvent	yield (%) ^b
1	NaOMe	EtOH	57
2	KOMe	EtOH	62
3	NaO ^t Bu	EtOH	78
4	KO ^t Bu	EtOH	84
5 ^c	KO ^t Bu	DMF	71
6 ^c	KO ^t Bu	DMSO	62
7	NaH	toluene	33
8	KOH	ethanol	24
9	-	ethanol	0

^aTwo equivalents of the base was used. ^bYield of the isolated product. ^cReaction was performed at 120 °C.

induced cyclization reaction in high boiling solvents DMF and DMSO could not improve the yield of the product 3a (entries 5–6), and in the absence of the base, the reaction could not afford the product 3a (entry 9). The reaction of 1a and 2a with 1 equiv of the base KO^tBu, under the optimized reaction condition, provided less yield of 3a (76%). With the optimized reaction conditions in hand (Table 1, entry 4), we then explored the substrate scope of the reaction with some representative 1,5-dicarbonyls (1a–h) and 3-amino-1H-pyrazoles (2a–g) which are shown in Table 2. The base induced cyclization reaction of 1,5-dicarbonyl 1a with various 3-amino-1H-pyrazoles (2a–g) without substituents and with substituents such as methyl, *t*-butyl, phenyl and 4-fluorophenyl present in the pyrazole ring reacted smoothly to afford pyrazolo[1,5-*a*]pyrimidines 3a–f in 70–84% yields under the optimized reaction conditions. Similarly, the reaction of 1a and 5-(thiophen-2-yl)-3-amino-1H-pyrazole (2g) afforded pyrazolo[1,5-*a*]pyrimidine 3g in 67% yield. Next, the reaction of 5-methyl-3-amino-1H-pyrazole (2b) with various symmetrical and unsymmetrical 1,5-dicarbonyls with electron donating and electron-withdrawing groups present in the aromatic rings were studied to extend the substrate scope. Symmetrical 1,5-dicarbonyls with methyl and chloro groups present in the aromatic rings 1b–c reacted efficiently with 2b to give the corresponding substituted pyrazolo[1,5-*a*]pyrimidines 3h–i in good yields (77–80%). Similarly, the cyclization reactions of 2b with symmetrical 1,5-dicarbonyls such as 3-phenyl-1,5-di-*p*-tolylpentane-1,5-dione (1d), 3-phenyl-1,5-di-*p*-chlorophenylpentane-1,5-dione (1e), 1,5-diphenyl-3-(*p*-tolyl)pentane-1,5-dione (1f) and 3-(*p*-chlorophenyl)-1,5-diphenylpentane-1,5-dione (1g) proceeded very easily under the optimized condition to afford pyrazolo[1,5-*a*]pyrimidines 3j–m in 71–80% yields. In addition, the reaction of 1f with 2a afforded 78% yield of pyrazolo[1,5-*a*]pyrimidine 3n, whose single X-ray crystallography studies (Figure 2) confirmed the structure of compound 3.¹² The reaction of unsymmetrical 1-phenyl-3,5-di-*p*-tolylpentane-1,5-dione (1h) with 2b afforded a mixture of pyrazolo[1,5-*a*]pyrimidines 3l (41%) and 3h (32%) under the standard condition. Similarly, the condensation reaction of unsymmetrical 1-(4-chlorophenyl)-3-phenyl-5-(*p*-tolyl)pentane-1,5-dione (1i) with 2b afforded a mixture of

Table 2. Synthesis of Various Substituted Pyrazolo[1,5-*a*]pyrimidines^a

^aReaction conditions: 1,5-dicarbonyl (1.0 mmol), 3-amino pyrazole (1.0 mmol) and KO^tBu (2.0 mmol) in ethanol (3.0 mL) was heated at 120 °C for 12 h; isolated yields.

pyrazolo[1,5-*a*]pyrimidines 3j (44%) and 3k (25%) under the standard condition. This result indicated that phenyl ring substituted with electron withdrawing group eliminated

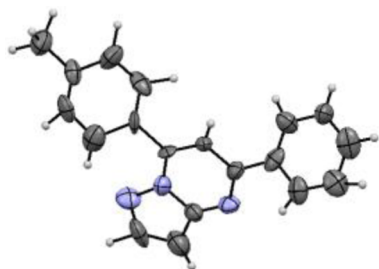
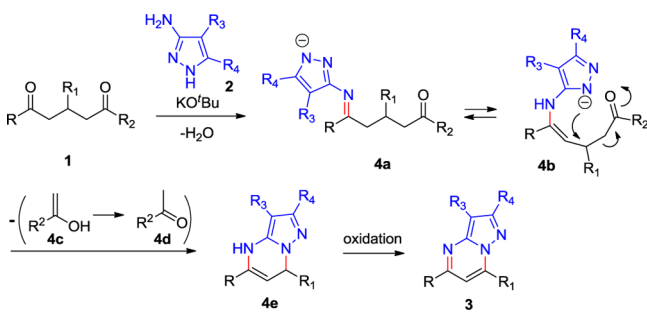


Figure 2. X-ray crystal structure of **3n**.

preferentially from the 1,5-dicarbonyl compound providing better yield of compound **3j**. The starting 1,5-dicarbonyl compounds were already used by different research groups for the construction of various important scaffolds^{13a,b} and they were easily prepared by refluxing a mixture of acetophenone with chalcone following known procedure.^{13c}

A probable mechanism for the formation of compound **3** is shown in Scheme 1. First, condensation of ketone **1** with 3-

Scheme 1. Proposed Reaction Mechanism



amino pyrazole **2**, followed by deprotonation of the formed imine derivative in the presence of base potassium *tert*-butoxide, generates the pyrazolide anion **4a**. Tautomerization of the imine **4a** to the enamine **4b**, followed by intramolecular nucleophilic attack of pyrazolide anion on the benzylic/allylic carbon (C3), affords intermediate **4e** by elimination of one molecule of acetophenone derivative **4d**. Finally, aerial oxidation of the resulting dihydropyrazolo[1,5-*a*]pyrimidine **4e** affords compound **3**.¹⁴ To prove the mechanism, the GC-MS data of the crude reaction mixture of **1d** and **2b** was recorded (see Supporting Information), as well as, the side product of the reaction was isolated. The side product was found to be 4-methylacetophenone, which proved the proposed mechanism for the formation of **3** from intermediate **4b** (Scheme 1). The substrate **1d** on treatment with potassium *tert*-butoxide alone under standard condition did not provide the base promoted eliminated product 4-methylacetophenone which further suggested the reaction mechanism depicted in Scheme 1.

In conclusion, we have developed a novel base induced reaction of 1,3,5-trisubstituted pentane-1,5-diones with substituted pyrazoles for the construction of poly substituted pyrazolo[1,5-*a*]pyrimidines via C-C/C-N bond cleavage reaction. The methodology has advantages such as wide substrate scope, easily available starting materials, high yields, and simple reaction procedure.

EXPERIMENTAL SECTION

General Information. Melting points were uncorrected. IR spectra were recorded using chloroform. NMR spectra were recorded on a 300 or 500 MHz spectrometer using tetramethylsilane (TMS) as an internal standard. Mass spectra were recorded on GCMS instrument. All the commercially available reagents were used as received. All experiments were monitored by thin layer chromatography (TLC). TLC was performed on precoated silica gel plates. Column chromatography was performed on silica gel (100–200 mesh).

General Procedure for the Synthesis of Pyrazolo[1,5-*a*]pyrimidines Derivatives **3a–n.** A mixture of 1,5-diketone **1** (1.0 mmol), 3-amino-1*H*-pyrazole **2** (1.0 mmol), and KO^tBu (2.0 mmol) in anhydrous ethanol was refluxed for 12 h. After completion of the reaction, the solvent was removed from the reaction mixture, water was added into it, and then it was extracted with ethyl acetate. The ethyl acetate layer was then washed with brine and water. Finally, it was dried over anhydrous Na₂SO₄, and the solvent was removed under vacuo. The crude product obtained was purified by column chromatography over silica gel (100–200 mesh) using EtOAc/hexane as the eluant.

5,7-Diphenylpyrazolo[1,5-*a*]pyrimidine (3a**).** Yellow solid (228 mg, 84%); mp: 88–90 °C. ¹H NMR (CDCl₃, 300 MHz) δ 6.81 (s, 1H), 7.34 (s, 1H), 7.51 (d, *J* = 6.5 Hz, 2H), 7.56–7.59 (m, 3H), 8.06 (d, *J* = 3.4 Hz, 2H), 8.11–8.17 (m, 4H); ¹³C NMR (CDCl₃, 75 MHz) δ 97.2, 105.2, 127.3, 128.4, 128.7, 128.9, 129.3, 130.3, 130.9, 131.6, 137.5, 145.2, 146.9, 156.2; IR (CHCl₃, cm⁻¹): 760, 1028, 1222, 1377, 1491, 1549, 1607, 2924. MS (EI, *m/z*): 271 [M⁺]. Anal. Calcd for C₁₈H₁₃N₃: C, 79.68; H, 4.83; N, 15.49. Found: C, 79.55; H, 4.99; N, 15.68.

2-Methyl-5,7-diphenylpyrazolo[1,5-*a*]pyrimidine (3b**).** Yellow solid (231 mg, 81%); mp: 115–117 °C. ¹H NMR (CDCl₃, 500 MHz) δ 2.53 (s, 3H), 6.57 (s, 1H), 7.48–7.50 (m, 6H), 7.55 (d, *J* = 4.2 Hz, 2H), 8.06–8.10 (m, 3H); ¹³C NMR (CDCl₃, 125 MHz) δ 14.8, 96.4, 104.3, 127.1, 128.6, 128.8, 129.2, 130.0, 130.8, 131.6, 137.6, 146.1, 150.6, 155.4, 155.7; IR (CHCl₃, cm⁻¹): 771, 1017, 1218, 1373, 1490, 1554, 1608, 2924. MS (EI, *m/z*): 285 [M⁺]. Anal. Calcd for C₁₉H₁₅N₃: C, 79.98; H, 5.30; N, 14.73. Found: C, 79.81; H, 5.59; N, 14.89.

2-*tert*-Butyl-5,7-diphenylpyrazolo[1,5-*a*]pyrimidine (3c**).** Gum (242 mg, 74%); ¹H NMR (CDCl₃, 500 MHz) δ 1.43 (s, 9H), 6.65 (s, 1H), 7.45–7.48 (m, 4H), 7.50–7.54 (m, 3H), 8.09 (d, *J* = 4.8 Hz, 2H), 8.20 (d, *J* = 5.7 Hz, 2H); ¹³C NMR (CDCl₃, 125 MHz) δ 30.4, 32.9, 93.0, 104.0, 127.1, 128.3, 128.8, 129.4, 129.9, 130.7, 131.6, 137.8, 145.9, 150.4, 155.4, 168.0; IR (CHCl₃, cm⁻¹): 762, 1017, 1239, 1490, 1551, 1606, 2960. MS (EI, *m/z*): 327 [M⁺]. Anal. Calcd for C₂₂H₂₁N₃: C, 80.70; H, 6.46; N, 12.83. Found: C, 80.96; H, 6.61; N, 12.55.

3,5,7-Triphenylpyrazolo[1,5-*a*]pyrimidine (3d**).** Yellow solid (250 mg, 72%); mp: 175–177 °C. ¹H NMR (CDCl₃, 500 MHz) δ 7.40 (s, 1H), 7.50–7.56 (m, 7H), 7.60 (d, *J* = 6.6 Hz, 2H), 8.06 (d, *J* = 5.9 Hz, 2H), 8.21–8.25 (m, 4H), 8.49 (s, 1H); ¹³C NMR (CDCl₃, 125 MHz) δ 105.1, 110.6, 126.0, 126.3, 127.3, 128.6, 128.8, 129.2, 130.3, 130.9, 132.3, 137.3, 142.9, 145.9, 146.9, 155.8; IR (CHCl₃, cm⁻¹): 691, 761, 1028, 1189, 1377, 1491, 1562, 1607, 2923. MS (EI, *m/z*): 347 [M⁺]. Anal. Calcd for C₂₄H₁₇N₃: C, 82.97; H, 4.93; N, 12.10. Found: C, 82.68; H, 4.75; N, 12.39.

3-(4-Fluorophenyl)-5,7-diphenylpyrazolo[1,5-*a*]pyrimidine (3e**).** Yellow solid (256 mg, 70%); mp: 159–161 °C. ¹H NMR (CDCl₃, 500 MHz) δ 7.19 (d, *J* = 5.4 Hz, 2H), 7.41 (s, 1H), 7.55 (d, *J* = 8.4 Hz, 4H), 7.57 (s, 1H), 7.60 (d, *J* = 3.3 Hz, 1H), 8.07–8.24 (m, 6H), 8.44 (s, 1H); ¹³C NMR (CDCl₃, 125 MHz) δ 105.1, 109.8, 115.4, 115.6, 127.2, 127.7, 127.8, 128.7, 128.9, 129.2, 130.4, 130.9, 131.3, 137.2, 142.6, 146.9, 155.9; IR (CHCl₃, cm⁻¹): 750, 835, 1227, 1492, 1565, 1613, 2924. MS (EI, *m/z*): 365 [M⁺]. Anal. Calcd for C₂₄H₁₆FN₃: C, 78.89; H, 4.41; N, 11.50. Found: C, 78.90; H, 4.62; N, 11.71.

2,5,7-Triphenylpyrazolo[1,5-*a*]pyrimidine (3f**).** Yellow solid (246 mg, 71%); mp 161–162 °C. ¹H NMR (CDCl₃, 500 MHz) δ 7.10 (s, 1H), 7.36 (s, 1H), 7.45–7.61 (m, 8H), 8.02 (d, *J* = 3.1 Hz, 1H) 8.14–8.22 (m, 6H); ¹³C NMR (CDCl₃, 125 MHz) δ 93.7, 104.9, 126.5,

127.2, 128.2, 128.5, 128.6, 128.8, 129.4, 130.2, 130.9, 131.4, 137.5, 146.3, 150.9, 156.0, 156.2; IR (CHCl₃, cm⁻¹): 759, 845, 1027, 1233, 1490, 1552, 1607, 2926. MS (EI, *m/z*): 347 [M⁺]. Anal. Calcd for C₂₄H₁₇N₃: C, 82.97; H, 4.93; N, 12.10. Found: C, 82.78; H, 4.81; N, 12.34.

5,7-Diphenyl-2-(thiophen-2-yl)pyrazolo[1,5-*a*]pyrimidine (3g). Yellow solid (237 mg, 67%); mp: 182–184 °C. ¹H NMR (CDCl₃, 500 MHz) δ 6.96 (s, 1H), 7.09–7.58 (m, 10H), 8.12 (d, *J* = 5.4 Hz, 2H), 8.18 (d, *J* = 4.8 Hz, 2H); ¹³C NMR (CDCl₃, 125 MHz) δ 93.5, 105.0, 125.7, 126.2, 127.2, 127.6, 128.5, 128.8, 129.4, 130.2, 130.9, 137.4, 146.2, 150.9, 151.5, 156.2; IR (CHCl₃, cm⁻¹): 762, 1017, 1218, 1490, 1565, 1606, 2923. MS (EI, *m/z*): 353 [M⁺]. Anal. Calcd for C₂₂H₁₅N₃S: C, 74.76; H, 4.28; N, 11.89. Found: C, 74.59; H, 4.55; N, 11.67.

2-Methyl-5,7-di-*p*-tolylpyrazolo[1,5-*a*]pyrimidine (3h). Gum (251 mg, 81%); ¹H NMR (CDCl₃, 500 MHz) δ 2.42 (s, 3H), 2.46 (s, 3H), 2.52 (s, 3H), 6.54 (s, 1H), 7.21 (s, 1H), 7.31 (d, *J* = 4.8 Hz, 2H), 7.37 (d, *J* = 4.5 Hz, 2H), 7.86 (d, *J* = 4.9 Hz, 2H), 7.97–7.99 (m, 2H); ¹³C NMR (CDCl₃, 125 MHz) δ 14.8, 21.5, 96.1, 103.8, 127.0, 129.3, 129.5, 134.6, 140.3, 141.2, 155.2, 155.7; IR (CHCl₃, cm⁻¹): 762, 826, 1176, 1478, 1610, 2937. MS (EI, *m/z*): 313 [M⁺]. Anal. Calcd for C₂₁H₁₉N₃: C, 80.48; H, 6.11; N, 13.41. Found: C, 80.16; H, 6.32; N, 13.66.

5,7-Bis(4-chlorophenyl)-2-methylpyrazolo[1,5-*a*]pyrimidine (3i). Yellow solid (272 mg, 77%); 172–175 °C. ¹H NMR (CDCl₃, 500 MHz) δ 2.54 (s, 3H), 6.59 (s, 1H), 7.19 (s, 1H), 7.49 (d, *J* = 4.8 Hz, 4H), 7.56 (d, *J* = 5.1 Hz, 4H), 8.06 (d, *J* = 5.9 Hz, 2H); ¹³C NMR (CDCl₃, 125 MHz) δ 29.6, 96.7, 103.7, 128.4, 129.0, 129.8, 135.8, 136.4, 137.1, 145.0, 150.4, 154.3, 155.7; IR (CHCl₃, cm⁻¹): 774, 817, 1013, 1090, 1486, 1593, 1607, 2924. MS (EI, *m/z*): 353 [M⁺]. Anal. Calcd for C₁₉H₁₃Cl₂N₃: C, 64.42; H, 3.70; N, 11.86. Found: C, 64.79; H, 3.84; N, 11.62.

2-Methyl-7-phenyl-5-*p*-tolylpyrazolo[1,5-*a*]pyrimidine (3j). Brown gum (227 mg, 76%); ¹H NMR (CDCl₃, 500 MHz) δ 2.43 (s, 3H), 2.53 (s, 3H), 6.55 (s, 1H), 7.22 (s, 1H), 7.31 (d, *J* = 7.8 Hz, 2H), 7.55–7.58 (m, 2H), 8.01 (d, *J* = 4.2 Hz, 2H), 8.06–8.09 (m, 3H); ¹³C NMR (CDCl₃, 125 MHz) δ 14.8, 21.3, 96.2, 104.2, 127.0, 128.6, 129.2, 129.5, 130.8, 131.6, 134.7, 140.4, 146.1, 150.5, 155.3, 155.7; IR (CHCl₃, cm⁻¹): 765, 817, 1180, 1492, 1606, 2924. MS (EI, *m/z*): 299 [M⁺]. Anal. Calcd for C₂₀H₁₇N₃: C, 80.24; H, 5.72; N, 14.04. Found: C, 80.02; H, 5.84; N, 13.98.

5-(4-Chlorophenyl)-2-methyl-7-phenylpyrazolo[1,5-*a*]pyrimidine (3k). Brown solid (230 mg, 72%); mp: 132–134 °C. ¹H NMR (CDCl₃, 500 MHz) δ 2.54 (s, 3H), 6.57 (s, 1H), 7.20 (s, 1H), 7.49 (s, 1H), 7.56–7.58 (m, 3H), 8.05–8.08 (m, 5H); ¹³C NMR (CDCl₃, 125 MHz) δ 14.9, 96.6, 104.0, 128.5, 128.7, 129.1, 129.3, 131.0, 131.5, 136.4, 146.4, 154.5, 155.7; IR (CHCl₃, cm⁻¹): 764, 827, 1013, 1488, 1556, 1594, 2924. MS (EI, *m/z*): 319 [M⁺]. Anal. Calcd for C₁₉H₁₄ClN₃: C, 71.36; H, 4.41; N, 13.14. Found: C, 71.59; H, 4.65; N, 13.01.

2-Methyl-5-phenyl-7-*p*-tolylpyrazolo[1,5-*a*]pyrimidine (3l). Brown solid (239 mg, 80%); mp: 109–111 °C. ¹H NMR (CDCl₃, 500 MHz) δ 2.46 (s, 3H), 2.53 (s, 3H), 6.58 (s, 1H), 7.21 (s, 1H), 7.24 (s, 2H), 7.49–7.52 (m, 3H), 7.99–8.11 (m, 4H); ¹³C NMR (CDCl₃, 125 MHz) δ 14.9, 21.6, 96.2, 96.4, 103.9, 104.1, 127.3, 128.9, 129.2, 129.4, 129.6, 130.1, 141.3, 146.4, 155.3, 155.8; IR (CHCl₃, cm⁻¹): 772, 1019, 1463, 1596, 1661, 2922. MS (EI, *m/z*): 299 [M⁺]. Anal. Calcd for C₂₀H₁₇N₃: C, 80.24; H, 5.72; N, 14.04. Found: C, 80.46; H, 5.44; N, 14.27.

7-(4-Chlorophenyl)-2-methyl-5-phenylpyrazolo[1,5-*a*]pyrimidine (3m). Gum (227 mg, 71%); ¹H NMR (CDCl₃, 500 MHz) δ 2.54 (s, 3H), 6.58 (s, 1H), 7.51 (s, 2H), 7.56–7.59 (m, 3H), 8.08–8.12 (m, 5H); ¹³C NMR (CDCl₃, 125 MHz) δ 14.9, 96.5, 104.5, 127.3, 128.7, 128.9, 129.3, 130.1, 130.9, 139.0, 143.9, 154.8, 155.0; IR (CHCl₃, cm⁻¹): 763, 1028, 1217, 1373, 1490, 1554, 1607, 2923. MS (EI, *m/z*): 319 [M⁺]. Anal. Calcd for C₁₉H₁₄ClN₃: C, 71.36; H, 4.41; N, 13.14. Found: C, 71.12; H, 4.69; N, 13.35.

5-Phenyl-7-*p*-tolylpyrazolo[1,5-*a*]pyrimidine (3n). White solid (222 mg, 78%); mp: 110–112 °C. ¹H NMR (CDCl₃, 300 MHz) δ 2.47 (s, 3H), 6.79 (s, 1H), 7.33 (s, 1H), 7.35–7.55 (m, 4H), 7.96 (d, *J* = 3.4 Hz, 2H), 8.11–8.16 (m, 4H); ¹³C NMR (CDCl₃, 75 MHz) δ

21.6, 97.1, 104.9, 127.3, 128.6, 128.9, 129.2, 129.4, 130.3, 137.6, 141.4, 145.1, 146.9, 156.2, 161.3; IR (CHCl₃, cm⁻¹): 2924, 1610, 1550, 1369, 1220, 1028, 761. MS (EI, *m/z*): 285 [M⁺]. Anal. Calcd for C₁₉H₁₅N₃: C, 79.98; H, 5.30; N, 14.73. Found: C, 79.65; H, 4.91; N, 14.55.

■ ASSOCIATED CONTENT

Supporting Information

Copies of ¹H NMR and ¹³C NMR spectra for 3a–n and X-ray crystallographic data (CIF file) for 3n. The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.joc.5b00933.

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Notes

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